



6340 Lake Worth Blvd #403, Fort Worth, TX 76135 Phone 817-236-2010 Fax 817-236-2008
Email: info@thestowefoundation.org

Surface Science and X-ray Spectroscopy Group

[Stanford Synchrotron Radiation Laboratory](#)

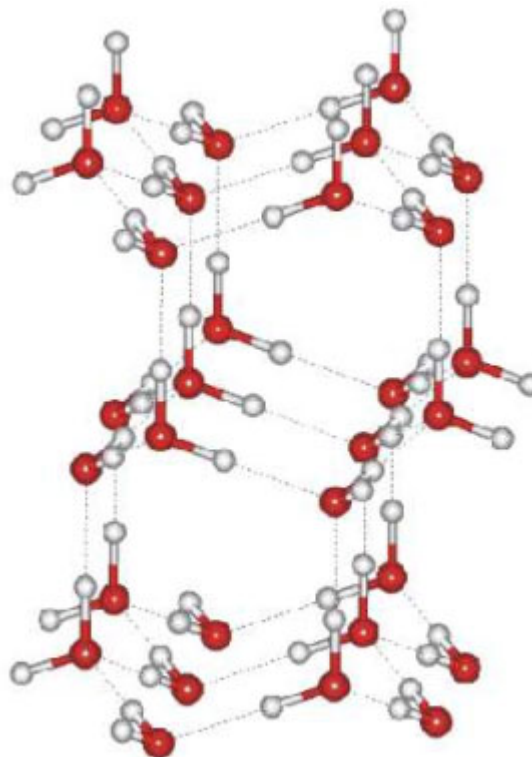
[Stanford Linear Accelerator Center](#)

Current Research Projects

Local Structure of liquid Water

Water molecules exhibit two types of interactions in the liquid and solid phases: strong covalent bonds within the molecule (O-H bonds) and relatively weak hydrogen bonds between them. In ice each molecule is coordinated by four neighboring waters through hydrogen bonds (two hydrogen bonds on the oxygen atom and one on each hydrogen). Although liquid water primarily is expected to have the same coordination environment, experimental (vibrational spectroscopy, neutron and x-ray diffraction) and theoretical (molecular dynamics) studies indicate that liquid water contains a fraction of molecules with broken hydrogen bonds. Still, despite intense investigations the local structure of liquid water is not understood to date.

The hydrogen bond in liquid water holds the key to the structure and properties of water, with implications for chemical, biological and geological processes. The dynamical motion of atoms at the picosecond (10⁻¹² s) timescale causes the hydrogen bonds to break and reform resulting in a statistical distribution of different coordinations. What is the nature of the hydrogen bond in terms of changes in the electronic structure? Can we understand this on a molecular orbital level?



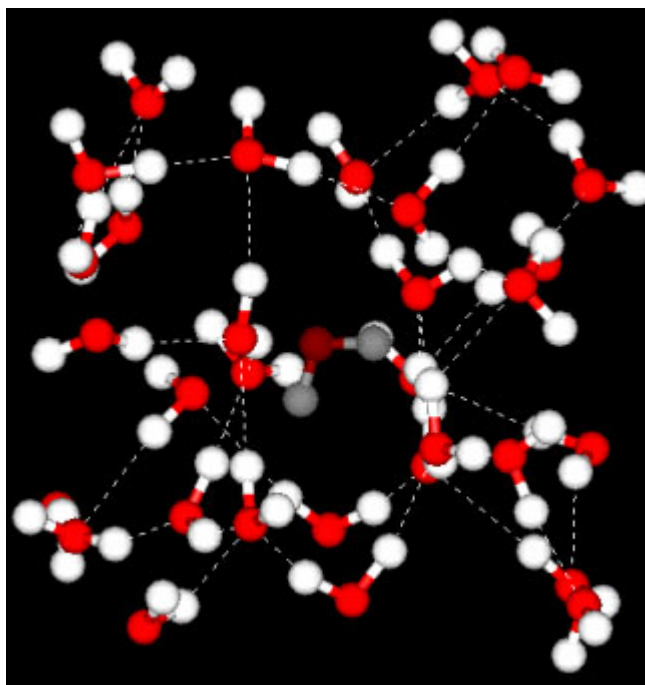
ordered structure of ice
(snapshot from a molecular dynamics simulation)



6340 Lake Worth Blvd #403, Fort Worth, TX 76135 Phone 817-236-2010 Fax 817-236-2008
Email: info@thestowefoundation.org

With x-ray absorption spectroscopy (at the oxygen K-edge) as a probe of the empty electronic states, we can locally characterize the nature of hydrogen bonding through the large effect of chemical bonding on the valence orbitals of the water molecules. We can identify different local configurations in the liquid and help solving the longstanding water puzzle.

Regular coordination in ice (above) and disordered structure in liquid water (right). Oxygen atoms are shown in red, hydrogens in white.



disordered structure of water
(snapshot from a molecular dynamics simulation)

Additional articles:

[Science Journal: The structure of water isn't certain after all](#)

Friday, March 10, 2006

By Sharon Begley, The Wall Street Journal